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STUDY ON THE SYNTHESIS OF NOVEL QUINONE-FUSED HETEROCYCLIC DERIVATIVES USING DOMINO REACTION AND THEIR BIOLOGICAL EVALUATION

Major : Organic Chemistry Code : 9 44 01 14

SUMMARY OF CHEMISTRY DOCTORAL THESIS

The work was completed at: Graduate University of Science and Technology

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INTRODUCTION

1. The urgency of the thesis

Quinone compounds are found in abundance in nature. Its synthetic, semi-synthetic and natural derivatives play an important role in many fields of chemistry and biology due to their interesting biological activities. They have attracted attention of chemists for their cytotoxicity through protein or DNA alkylation and the influence on redox processes by their semiquinone radicals as free radical carriers. These activities promoted inflammatory processes, oxidized DNA and induced cancer cell death. Several quinone-based drugs such as doxorubicin and mitomycin C have been used clinically for cancer chemotherapy. Therefore, scientists are continuing to search and develop new compounds with higher anti-tumor potential based on the quinone framework.

The quinone-fused heterocyclic compounds containing oxygen and nitrogen account for a large proportion in this class of substances, typically naphtho[2,3-*b*]furan-4,9-dione, naphtho[2,3-*c*]chromene-7,12-dione, aza-anthraquinone, benzo [*f*]indole-4,9-dione, ... Scientific reports showed that they have many activities such as antifungal, antibacterial, anti-inflammatory, anti-cancer, anti-HIV, anti-malarial, anti-parasitic... These compounds have been successfully synthesized from various methods such as domino reaction, Michael reaction, Diels-Alder cyclic addition reaction, Claisen reaction, metal catalyzed reaction, redox reaction, coupling reaction, etc. Among these methods, the domino reactions showed many advantages such as increasing the complexity of the product structure due to the formation of many new bonds, high stereoselectivity, and no need to isolate intermediate compounds.

Therefore, we conducted the thesis "Study on the synthesis of novel quinone-fused heterocyclic derivatives using domino reaction and their biological evaluation" in order to synthesize new quinone-fused heterocyclic compounds containing oxygen and nitrogen heteroatoms, with potential bioactivities as a basis for further studies. This thesis is scientific and practical significance. The objective of the study is to synthesize some new heterocyclic derivatives of benzo[g]chromene-5,10-dione, podophyllotoxin-naphthoquinone, benzo[a]pyridazine[3,4-c] phenazine through multicomponent domino reactions, and evaluate cytotoxic activity against cancer cell lines of some synthetic compounds.

2. Research objectives of the thesis

- Research and application domino reaction to synthesize new quinone derivatives containing oxygen and nitrogen heteroatoms.

- Cytotoxic evaluation of synthesized compounds.

3. Research content of the thesis

- Synthesis of 3-benzoyl-4*H*-benzo[*g*]chromene-5,10-dione compounds from the domino reaction.

- Synthesis of podophyllotoxin-naphthoquinone compounds from domino reactions.

- Synthesis of *N*-arylated-dihydrobenzo[g]quinoline-5,10-dione compounds from the domino reaction.

- Synthesis of benzo[*a*]pyridazino[3,4-*c*]phenazine compounds from the domino reaction.

- Cytotoxic evaluation of some synthesized compounds on cancer cell lines: epidermoid carcinoma (KB), hepatoma carcinoma (HepG2), non-small lung (SK-Lu-1 or A549) and breast (MCF7) cancer cell lines .

CHAPTER 1. OVERVIEW

Chapter 1 consisting of 24 pages, presents a literature review on quinone fused heterocyclic compounds containing oxygen and nitrogen and their biological activities; domino reaction and studies on applying domino reaction in the synthesis of quinone fused heterocyclic compounds.

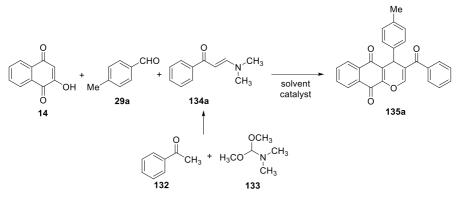
CHAPTER 2. RESEARCH METHODS AND EXPERIENCE

Chapter 2 consisting of 49 pages, presents the research methods, synthesis process, purification, reaction yield, physical properties of synthesized compounds such as morphology, color, melting point and their spectral data (IR, ¹H NMR, ¹³C NMR, HSQC, HMBC and HRMS).

CHAPTER 3. RESULTS AND DISCUSSION

3.1. Synthesis results of 3-benzoyl-4*H*-benzo[*g*]chromene-5,10-dione derivatives

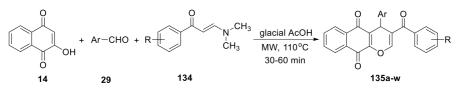
Initially, the reaction of 2-hydroxy-1,4-naphthoquinone (14), 4methylbenzaldehyde (29a) and enaminone (133a) (formed by the condensation of N,N-dimethyformamide dimethyl acetal with acetophenone in toluene at reflux for 1 h) according to scheme 3.1. was chosen for developing the optimal reaction conditions.



Scheme 1. Synthesis of compound 135a

This reaction was carried out under MW in different solvents (CH₃CN, t-BuOH, EtOH, dioxane, toluene, AcOH) at a range of temperatures (80–120°C) in the presence or absence of catalyst. Experimental results showed that the synthesis of compound **135a** has the highest yield when heating at 110°C in glacial acetic acid under microwave irradiation for 40 min.

Using the optimized reaction conditions, a series of 3-benzoyl-4*H*-benzo[*g*]chromene-5,10-dione derivatives **135a-w** with different substitutions on the aryl groups were synthesized (scheme 3.2, table 3.2). The reaction yield of compounds is from 60 to 88%. The structures of all synthesized compounds **135a-w** were elucidated using IR, NMR and HRMS spectroscopic techniques.



Scheme 3.2. Synthesis of compounds 135a-w

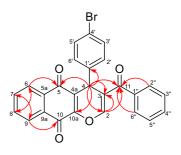
Entry	Product	Ar-	R	Yeild (%)
1	135a	$4-MeC_6H_4$	Н	84%
2	135b	4-MeOC ₆ H ₄	Н	63%
3	135c	$2-FC_6H_4$	Н	72%
4	135d	$4-BrC_6H_4$	Н	88%
5	135e	$4-CF_3C_6H_4$	Н	68%
6	135f	3-MeO-4- OHC ₆ H ₃	Н	66%
7	135g	C_6H_5	4-Me	61%
8	135h	$4-MeC_6H_4$	4-Me	82%
9	135i	3-MeOC ₆ H ₄	4-Me	69%
10	135j	$4-ClC_6H_4$	4-Me	75%
11	135k	$4-BrC_6H_4$	4-Me	78%
12	1351	C_6H_5	4-F	60%
13	135m	$4-MeC_6H_4$	4-F	83%

14	135n		4-F	73%
15	1350	2-naphthyl	4-F	65%
16	135p	$3-BrC_6H_4$	4-F	75%
17	135q	$4-BrC_6H_4$	4-F	78%
18	135r	$4-NO_2C_6H_4$	4-F	68%
19	135s	C_6H_5	3-OH	63%
20	135t	$4-MeC_6H_4$	3-OH	77%
21	135u	$2-FC_6H_4$	3-OH	60%
22	135v	$4-BrC_6H_4$	3-OH	65%
23	135w	$4-NO_2C_6H_4$	3-OH	64%

On the ¹H NMR spectrum of compound **135d**, there is a singlet signal at 5,40 ppm (1H, s, H-4) of the H-4 proton and a singlet signal at 7,51 ppm (1H, s, H-2) of the cyclic H-2 proton chromene. Signal pairs at position $\delta_{\rm H}$ = 8,16 – 8,15 ppm (1H, m, H-9), 8,04 – 8,02 ppm (1H, m, H-6), 7,75 – 7,74 ppm (2H, m, H-7, H-8) is that of the four naphthoquinone cyclic protons. In addition, the signal of 9 protons of two aromatic rings appeared to overlap in the low-field region from 7,58 to 7,35 ppm.

¹³C NMR spectrum of compound **135d** appeared enough signals of 26 carbon atoms in the molecule, including typical signals of 3 carbonyl carbon atoms at 193,1 ppm (C-11), 182,9 ppm (C- 5), 177,8 ppm (C-10). In

the higher field is the signal of the carbon atoms bonded to the oxygen atom in the chromene ring at 150,5 ppm (C-2) and 149,1 ppm (C-10a). The C-4 carbon atom gives a characteristic signal at 34,2 ppm, the remaining signals of the naphthoquinone and aromatic ring carbons appear from 141,3 to 120,3 ppm.



The results of HMBC and HSQC spectral analysis of compound **135d** presented in table 3.3 confirmed the expected structural formula and fully attributed the proton and carbon signals.

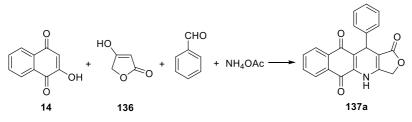
Interactions on the				
No.	C (ppm)	δ _H (ppm) (mult,, J (Hz))	HMBC	
110.	C (ppm)	0 H (pp m) (mait, 3 (112))	$(\mathbf{H} \rightarrow \mathbf{C})$	
2	150.1	7 51 (1H a)		
-	150,1	7,51 (1H, s)	3, 10a, 11	
3	120,3	-	-	
4	34,2	5,40 (1H, s)	2, 3, 4a, 5, 10a, 11	
			1', 2', 6'	
4a	123,7	-	-	
5	182,9	-	-	
5a	131,7	-	-	
6	126,7	8,04-8,02 (1H, m)	5, 8	
7	134,7	7,75-7,74 (1H, m)	5a, 9	
8	134,0	7,75-7,74 (1H, m)	6	
9	126,6	8,16-8,15 (1H, m)	7, 10	
9a	130,6	-	-	
10	177,8	-	-	
10a	149,1	-	-	
11	193,1	-	-	
1'	141,3	-	-	
2'	130,5	7,35 (1H, d, 8,4 Hz)	4', 6'	
3'	131,9	7,42 (1H, d, 8,4 Hz)	1', 4'	
4'	121,6	-	-	
5'	131,9	7,42 (1H, d, 8,4 Hz)	1' 4'	
6'	130,5	7,35 (1H, d, 8,4 Hz)	2', 4'	

Table 3.3. NMR spectral data of compound 135d

1"	137,5	-	-
2"	128,8	7,58 (1H, dd, 1,8 Hz, 8,4 Hz)	4", 6", 11
3"	128,6	7,44 (1H, d, 7,8 Hz)	1''
4"	132,5	7,55 (1H, t, 7,2 Hz)	3', 5'
5"	128,6	7,44 (1H, d, 7,8 Hz)	1"
6''	128,8	7,58 (1H, dd, 1,8 Hz, 8,4 Hz)	2", 4", 11

On the high-resolution mass spectrometry of compound **135d**, fragments m/z $[M-H]^-$ 469,0097 and 471,0179 were found which match the theoretically calculated mass for the molecular formula $[C_{26}H_{14}BrO_4]^-$ (469,0075 and 471,0055). The structures of other compounds were similarly elucidated.

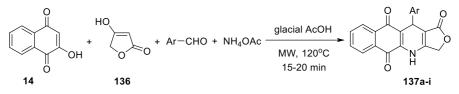
3.2. Synthesis of podophyllotoxin-naphthoquinone derivatives



Scheme 3.4. Synthesis of compound 137a

11-phenyl-4,11-dihydrobenzo[g]furo[3,4-b] The of synthesis quinoline-1,5,10(3*H*)-trione 137a starting from 2-hydroxy-1,4naphthoquinone (14) (1 mmol), tetronic acid (136) (1 mmol), benzaldehyde (1 mmol) and ammonium acetate (3 mmol) under microwave irradiation was chosen for the screening of reaction conditions (Scheme 3.4). This reaction was carried out in severent organic solvents such as ethanol, tbutanol and glacial acetic acid in the presence of molecular sieve as water absorption at range of 80-130°C. The results showed the reaction in glacial acetic acid at 120°C gave the highest yield (82%) of product 137a.

Based on the optimized reaction conditions (AcOH, 120°C), different substituted aromatic aldehydes were applied to this reaction to obtain a series of podophyllotoxin analogs **137a-i** (Scheme 3.5). The structures of all synthesized compounds **137a-i** were fully characterized by IR, NMR and HRMS spectroscopic techniques.

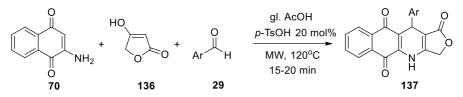


Scheme 3.5. Synthesis of compound **137a-i** from a four-component domino reaction

Entry	Product	Ar	Time (min)	Yield (%)
1	137 a	$2-FC_6H_4$	15	78
2	137b	$4-FC_6H_4$	15	80
3	137c	$4-BrC_6H_4$	15	78
4	137d	$4-ClC_6H_4$	15	81
5	137e	$4-CF_3C_6H_4$	15	80
6	137f	4-MeOC ₆ H ₄	20	85
7	137g	3-OMe-4- OHC ₆ H ₃	20	79
8	137h	1-naphthyl	20	79
9	137i	2-naphthyl	20	89

Table 3.4. Compounds 137a-i

In addition, compounds **137** were synthesized efficiently *via* microwave-assisted three-component reactions of 2-amino-1,4-naphthoquinone (**70**), tetronic acid (**136**), and (hetero) aromatic aldehydes in glacial acetic acid as a solvent and *p*-toluenesulfonic acid (*p*-TsOH, 20 mol%) as a catalyst (scheme 3.7).



Scheme 3.7. Synthesis of compound **137j-x** from a three-component domino reaction

Entry	Product	Ar	Road 1		Roa	nd 2
			Time	Yield	Time	Yield
			(min)	(%)	(min)	(%)
1	137j	3-OMeC ₆ H ₄	20	83	15	86
2	137k	$3-BrC_6H_4$	15	79	15	83
3	137 l	$3-NO_2C_6H_4$	25	67	15	75
4	137m	C_6H_5	15	82	15	84
5	137n	$4-\text{MeC}_6\text{H}_4$	20	88	15	88
6	1370	$2-NO_2C_6H_4$	26	65	15	73
7	137p	3-oxo-1,3-dihydro benzofuran-5-yl	20	82	15	85
8	137q	2-F-4-MeOC ₆ H ₃	20	39	20	71
9	137r	3-F-4-MeOC ₆ H ₃	20	42	20	74
10	137s	2,6-F ₂ -4-MeOC ₆ H ₂	25	37	25	70
11	137t	2-CF ₃ -4-MeOC ₆ H ₃	20	36	20	73
12	137u	2-F-4-OHC ₆ H ₃	20	17	20	72
13	137v	4-(4- fluorophenoxy)C ₆ H ₄	25	trace	25	69
14	137x	$4-CF_3OC_6H_4$	20	69	20	78

Table 3.6 . Compounds 137j-x

Experimental results showed that using the three-component domino reaction to synthesize podophyllotoxin-naphthoquinone

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compounds according to scheme 3.7 usually gives higher yields and shorter reaction time than using the four-component domino reaction according to scheme 3.5.

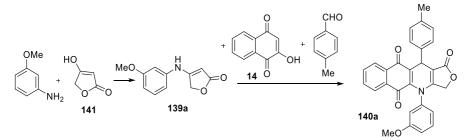
Then, we continued to using the three-component domino reaction according to the procedure as shown in diagram 3.7 to synthesize 8 podophyllotoxin-naphthoquinone derivatives **137aa-ah** with high yields (79 - 89%) (table 3.7). The structures of all synthesized compounds **135a-w** were elucidated using IR, NMR and HRMS spectroscopic techniques.

STT	Product	Ar	Reaction	Yield
			time	(%)
			(minutes)	
1	137 aa	$2-OHC_6H_4$	20	80
2	137ab	$4-OHC_6H_4$	20	79
3	137ac	$4-NO_2C_6H_4$	15	82
4	137ad	$4-CNC_6H_4$	15	89
5	137ae	3,4-(MeO) ₂ C ₆ H ₃	15	85
6	137af	3,4,5-(MeO) ₃ C ₆ H ₂	15	86
7	137ag	pyridin-3-yl	15	81
8	137ah	5-Br-pyridin-2-yl	15	83

Table 3.7. Compounds 137aa-ah

Thus, by using multicomponent domino reaction, we have successfully synthesized 31 podophyllotoxin-naphthoquinone compounds **137** from 2-amino-1,4-naphthoquinone or 2-hydroxy-1,4- naphthoquinone and ammonium acetate, aldehyde with tetronic acid in glacial acetic acid as a solvent under microwave irradiation.

3.3. Synthetic results of *N*-arylated-dihydrobenzo[g]quinoline-5,10dione derivatives In order to incorporate an aryl substituent onto the nitrogen atom of the podophyllotoxin-naphthoquinone framework, we synthesized 4-(3-methoxyphenyl)-11-(p-tolyl)-4,11-dihydrobenzo[g]furo[3,4-b]quinoline-1,5,10(3H)-trione (**140a**) by a three-component domino reaction as shown in scheme 3.9.



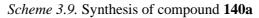


Table 3.8. Optimization of the reaction condition for the synthesis of

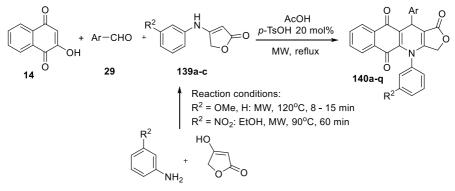
compound 140a					
Entry	Solvent	Catalyst (mol%)	Reaction time (min)	Yield (%)	
1	CH ₃ CN	_	90	0	
2	EtOH	_	90	0	
3	Toluene	_	90	0	
4	AcOH	_	90	0	
5	CH ₃ CN	TFA (10 mol%)	20	13	
6	EtOH	TFA (10 mol%)	20	17	
7	Toluene	TFA (10 mol%)	20	Trace	
8	AcOH	TFA (10 mol%)	20	43	
9	CH ₃ CN	<i>p</i> -TsOH (10 mol%)	20	14	
10	EtOH	<i>p</i> -TsOH (10 mol%)	20	20	
11	Toluene	<i>p</i> -TsOH (10 mol%)	20	11	
12	AcOH	<i>p</i> -TsOH (10 mol%)	20	57	

compound 140a

13	AcOH	<i>p</i> -TsOH (20 mol%)	20	83
14	AcOH	<i>p</i> -TsOH (30 mol%)	20	83

The reaction was investigated in different organic solvents and catalysts as indicated in table 3.8. The result revealed that the highest yield was obtained using glacial acetic acid as a solvent and 20 mol% of p-TsOH as a catalyst.

By using the optimised reaction conditions, we have successfully synthesized 17 compounds of *N*-arylated-dihydrobenzo[g]quinoline-5,10-dione **140a-q** from 2-hydroxy-1,4-naphthoquinone (**14**), aromatic aldehyde, aniline/aniline derivative and tetronic acid under microwave irradiation. Products **140a-q** were afforded in good yields (69–86%) within 20-60 min. The structure of 17 synthesized compounds **140** were fully characterized by IR, NMR and HRMS.



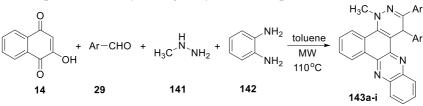
Scheme 3.10. Synthesis of compounds 140a-q

Entry	Product	Ar	R ²	Time	Yield
				(min)	(%)
1	140a	4-MeC ₆ H ₄	3-OMe	20	83
2	140b	$4-BrC_6H_4$	3-OMe	60	85
3	140c	$4-CF_3C_6H_4$	3-OMe	60	69
4	140d	C_6H_5	3-OMe	60	84

5	140e	4-MeC ₆ H ₄	3-NO ₂	20	80
6	140f	$4-BrC_6H_4$	3-NO ₂	40	83
7	140g	$4-CF_3C_6H_4$	3-NO ₂	40	80
8	140h	C_6H_5	3-NO ₂	40	86
9	140i	3-OMeC ₆ H ₄	3-NO ₂	40	78
10	140j	$3-NO_2C_6H_4$	3-NO ₂	60	81
11	140k	$4-MeC_6H_4$	Н	40	79
12	140 l	$4-BrC_6H_4$	Н	40	81
13	140m	$4-CF_3C_6H_4$	Н	40	80
14	140n	C_6H_5	Н	40	82
15	1400	3-OMeC ₆ H ₄	Н	40	79
16	140p	$3-NO_2C_6H_4$	Н	40	82
17	140q	$4-ClC_6H_4$	Н	40	81

3.4. Synthetic results of benzo[a]pyridazino[3,4-c]phenazine derivatives

In addition, we conducted a multicomponent domino reaction from 2-hydroxy-1,4-naphthoquinone (14), aromatic or heterocyclic aldehyde **29a-i**, methylhydrazine (132) and *o*-phenylenediamine (142) according to scheme 3.12 to synthesize a new heterocyclic compound containing phenazine and pyridazine-based heterocyclic radicals, which are structural subunits present in many biologically active compounds.



Scheme 3.12. Synthesis of compounds 143a-i

9 New benzo[a]pyridazino[3,4-c]phenazines **134a-i** were selectively obtained in 45-63% yield after purification by silica gel column chromatography. The proposed molecular structure of all synthesized

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compounds **143a-i** were fully characterized by IR, ¹H NMR, ¹³C NMR and HRMS.

Entry	Product	Ar	Yield (%)	
1	143a	C_6H_5	52	
2	143b	$4-ClC_6H_4$	55	
3	143c	$3-BrC_6H_4$	63	
4	143d	4-BrC ₆ H ₄	60	
5	143e	$4-NO_2C_6H_4$	48	
6	143f	$4-SO_2(CH_3)C_6H_4$	41	
7	143g	3-MeOC ₆ H ₄	57	
8	143h	4-MeOC ₆ H ₄	50	
9	143i	2-naphthyl	45	

Table 3.11. Compounds 143a-i

We also conducted the same reaction with *o*-substituted aromatic aldehydes (Ar = 2-FC₆H₄, 2-OMe C₆H₄, 2-NO₂-5-OHC₆H₃). However, the reaction of these aldehydes, 2-hydroxy-1,4-naphthoquinone (**14**), methylhydrazine (**141**) and *o*-phenylenediamine (**142**) didn't afford compounds **143**, probably due to steric hindrance between substituted group of aldehyde and nitrogen atom on phenazine core.

3.5. Cytotoxic activity

Podophyllotoxin-naphthoquinones **137**, *N*-arylated-dihydrobenzo[g] quinoline-5,10-diones **140** and benzo[a]pyridazino[3,4-c] phenazines **143** were selected to evaluate the cytotoxic activity against human cancer cell lines including mouth epidermal carcinoma KB, hepatoma carcinoma HepG2, lung cancer SK-Lu-1 or A549, and breast cancer MCF7, as well as human embryonic kidney cell line Hek-293. Ellipticine was used as positive control.

Table 3.13. Cytotoxicity of compounds 140a-q						
En	Com	IC 50 (µM)				
try	pound	KB	HepG2	A549	MCF7	Hek-293
1	140a	> 20	15,75 ± 1,08	> 20	> 20	-
2	140b	> 20	> 20	> 20	> 20	_
3	140c	> 20	> 20	> 20	> 20	-
4	140d	$14,\!97 \pm 1,\!11$	$2,22\pm0,22$	$11,\!26\pm0,\!65$	9,21 ± 0,33	$3,\!87\pm0,\!15$
5	140e	> 20	> 20	> 20	> 20	-
6	140f	> 20	> 20	> 20	> 20	-
7	140g	> 20	> 20	$13,\!48 \pm 0,\!94$	> 20	-
8	140h	$17{,}22\pm1{,}08$	12,91 ± 1,08	$6{,}13\pm0{,}32$	$10,\!39\pm0,\!47$	-
9	140i	> 20	1,23 ± 0,10	$16,17\pm1,01$	1,98 ± 0,30	$37,\!89 \pm 4,\!00$
10	140j	> 20	> 20	$7,\!75\pm0,\!29$	> 20	-
11	140k	> 20	> 20	> 20	> 20	-
12	140 l	$11,03 \pm 1,79$	$2,\!97\pm0,\!30$	$11,25 \pm 0,70$	$9,23 \pm 0,42$	$10,15 \pm 0,58$
13	140m	16,41 ± 1,03	9,12 ± 0,68	9,31 ± 0,72	$8,\!30\pm0,\!51$	$15,13 \pm 1,00$
14	140n	$8,\!79\pm0,\!36$	0,95 ± 0,12	$9{,}08 \pm 0{,}36$	0,91 ± 0,12	$10,33 \pm 1,54$
15	1400	$12,\!99 \pm 0,\!62$	$7,\!07\pm0,\!33$	$12,\!32\pm0,\!82$	$11,16 \pm 1,11$	$11,\!47 \pm 1,\!12$
16	140p	$5,79\pm0,22$	0,63 ± 0,02	$7{,}23\pm0{,}32$	0,81 ± 0,19	$4,\!96\pm0,\!36$
17	140q	> 20	> 20	> 20	> 20	-
Ellipticin		1.33	$1,\!33\pm0,\!20$	$1,\!42\pm0,\!20$	$2{,}55\pm0{,}20$	$2,\!35\pm0,\!20$

3.5.1. Cytotoxic activity of N-arylated-dihydrobenzo[g]quinoline-5,10dione compounds

According to concentrations of compounds inhibiting cell growth by 50% (IC₅₀ values) shown in table 3.13, compounds 140d, 140h, 140i and 1401-p exhibited low to moderate micromolar IC₅₀ values in the different cell lines evaluated (IC₅₀ < 18 μ M) with the exception of 140i, which exhibited no appreciable activity in KB cells. Interestingly,

compounds **140i**, **140n** and **140p** were particularly toxic in HepG2 and MCF7 cells with IC₅₀ < 2 μ M, that are similar to ellipticine. Besides that, compounds **140d** and **140l** expressed the high inhibitory activity against HepG2 cell line with IC₅₀ = 2,22 and 2,97 μ M, respectively. In addition, the selective cytotoxicity of the most potent cytotoxic compounds was also evaluated using non-cancerous human embryonic kidney (Hek-293) cells. In general, these compounds showed low to moderate toxic in the Hek-293 with IC₅₀ values ranging from 3,87 to 37,89 μ M.

3.5.2. Cytotoxic activity of benzo[a]pyridazino[3,4-c]phenazine compounds

En	Com	IC 50 (µM)				
try	pound	KB	HepG2	Lu-1	MCF7	Hek-293
1	14 3 a	$19,\!41 \pm 0,\!40$	$10,\!48 \pm 0,\!20$	$18,55 \pm 0,25$	$22,04 \pm 0,50$	> 100
2	143b	> 100	> 100	> 100	> 100	> 100
3	143c	> 100	> 100	> 100	> 100	> 100
4	143d	> 100	> 100	> 100	> 100	> 100
5	143e	> 100	> 100	> 100	> 100	> 100
6	143f	$62,\!48 \pm 1,\!80$	$44,\!68 \pm 1,\!03$	> 100	$62,74 \pm 1,17$	> 100
7	143g	$64,\!41 \pm 0,\!80$	$36,32 \pm 0,27$	$55,96 \pm 1,06$	$70,66 \pm 1,49$	> 100
8	143h	$16,70 \pm 0,34$	$14,\!62\pm0,\!15$	$18,25 \pm 0,23$	$21,83 \pm 0,19$	73.44 ± 1.34
9	143i	> 100	$81,\!42 \pm 0,\!94$	> 100	> 100	> 100
Ell	lipticin	1.26 ± 0.05	$1,34 \pm 0,04$	$1,\!83\pm0,\!05$	$2,\!48 \pm 0,\!05$	$6{,}58 \pm 0{,}04$

Table 3.2. Cytotoxicity of compounds **143a-i**

Results of the cytotoxicity test of benzo[a]pyridazino[3,4c]phenazine**143a-i**derivatives showed that compounds with -Cl, -Br, -NO₂groups (**143b-e**) were inactive against all four cancer cell lines, whereascompound**143f**having SO₂CH₃ group showed less active toward KB, $HepG2 and HCF7 cell lines with IC₅₀ values ranging from 44 to 63 <math>\mu$ M. The introduction of methoxy group to aryl ring (compounds **143g-h**) showed higher inhibitory activity in comparison with compounds **143b-f**. In particular, compound **143h** (Ar = 4-OMeC₆H₅) and compound **143a** (Ar = C₆H₅) displayed good cytotoxic effects on four tested cell lines with IC₅₀ values ranging from 10 to 23 μ M. Besides, compound **143i** having naphthyl group showed no activity against these cancer cell lines. All compounds **143a-i** exhibited lower toxic to the Hek-293 cells.

En	Comp	IC 50 (µM)				
try	ound	KB	HepG2	Lu-1	MCF7	Hek-293
1	137a	1,19 ± 0,01	0,42 ± 0,01	0,28 ± 0,01	> 2,50	10.30 ± 0.10
2	137b	2,16 ± 0,07	0,44 ± 0,01	$\textbf{0,}\textbf{28} \pm \textbf{0,}\textbf{03}$	> 2,50	5.35 ± 0.21
3	137c	2,14 ± 0,11	1,50 ± 0,01	0,45 ± 0,01	> 2,50	8.27 ± 0.14
4	137d	1,46 ± 0,01	1,35 ± 0,01	$0,\!50\pm0,\!01$	> 2,50	5.89 ± 0.12
5	137e	0,80 ± 0,01	1,24 ± 0,01	0,51 ± 0,01	> 2,50	7.59 ± 0.17
6	137f	0,88 ± 0,02	2,06 ± 0,02	$\textbf{0,88} \pm \textbf{0,05}$	> 2,50	8.17 ± 0.04
7	137g	1,90 ± 0,02	0,41 ± 0,01	0,41 ± 0,01	> 2,50	1.41 ± 0.01
8	137h	> 2,50	> 2,50	> 2,50	> 2,50	14.68 ± 0.12
9	137i	> 2,50	1,68 ± 0,07	$1,\!02\pm0,\!05$	> 2,50	7.61 ± 0.05
10	137j	1,23 ± 0,01	0,46 ± 0,01	< 0,040	> 2,50	9.35 ± 0.44
11	137k	1,71 ± 0,07	< 0,036	< 0,036	> 2,50	7.20 ± 0.05
12	137 l	1,16 ± 0,01	< 0,039	< 0,039	2,19 ± 0,02	1.47 ± 0.04
13	137m	1,54 ± 0,01	< 0,044	< 0,044	1,98 ± 0,04	1.34 ± 0.01
14	137n	> 2,50	2,24 ± 0,01	1,65 ± 0,03	> 2,50	$\textbf{21.59} \pm \textbf{0.40}$
15	1370	2,34 ± 0,26	> 2,50	$2,\!22\pm0,\!04$	> 2,50	$\textbf{9.95} \pm \textbf{0.01}$
16	137p	1,86 ± 0,02	0,47 ± 0,01	0,08 ± 0,01	> 2,50	4.78 ± 0.14
Ell	ipticin	1.75 ± 0.03	$1,66 \pm 0,03$	$1{,}54\pm0{,}03$	$1,\!58\pm0,\!03$	$1,\!69\pm0,\!04$

3.5.3. Cytotoxic activity of podophyllotoxin-naphthoquinone compounds Table 3.15. Cytotoxicity of compounds 137a-p

Ent	Comp	IC 50 (µM)					
ry	ound	KB	HepG2	A549	MCF7	Hek-293	
1	137q	$2{,}54\pm0{,}01$	3,13 ± 0,01	$4,\!42 \pm 0,\!01$	$2,\!65\pm0,\!01$	21.87 ± 0.01	
2	137r	2,36 ± 0,01	$3,24 \pm 0,01$	1,13 ± 0,07	$3,\!43 \pm 0,\!03$	10.04 ± 0.10	
3	137s	> 5,00	> 5,00	> 5,00	> 5,00	40.06 ± 0.21	
4	137t	> 5,00	> 5,00	> 5,00	> 5,00	21.91 ± 0.26	
5	137u	1,86 ± 0,03	> 5,00	1,82 ± 0,01	$2,\!69\pm0,\!03$	1.54 ± 0.14	
6	137v	0,60 ± 0,01	$1,05\pm0,03$	> 5,00	$2,\!12\pm0,\!04$	3.68 ± 0.02	
7	137x	1,11 ± 0,01	1,11 ± 0,03	1,39 ± 0,10	$2{,}59\pm0{,}03$	1.08 ± 0.01	
Elli	ipticin	1.53 ± 0.04	$1{,}50\pm0{,}03$	$1{,}58\pm0{,}03$	$1,\!83\pm0,\!07$	$6{,}33 \pm 0{,}04$	
	Table 3.17. Cytotoxicity of compounds 137aa - ah						
En	Comp			IC 50 (µM)			
try	ound	KB	HepG2	A549	MCF7	Hek-293	
1	137 aa	$0,\!57\pm0,\!02$	0,63 ± 0,02	0,43 ± 0,01	1,61 ± 0,03	2.03 ± 0.06	
2	137ab	> 2,50	> 2,50	1,59 ± 0,03	2,23 ± 0,06	6.46 ± 0.15	
3	137ac	> 2,50	> 2,50	$1,12 \pm 0,04$	> 2,50	7.17 ± 0.16	
4	137ad	> 2,50	> 2,50	$\textbf{2,03} \pm \textbf{0,05}$	> 2,50	> 20	
5	137ae	$0,\!52\pm0,\!02$	0,53 ± 0,02	1,10 ± 0,02	2,28 ± 0,07	2.23 ± 0.08	
6	137af	0,02 ± 0,01	0,02 ± 0,01	0,62 ± 0,02	0,12 ± 0,03	0.03 ± 0.01	
7	137ag	0,62 ± 0,01	> 2,50	0,61 ± 0,02	1,96 ± 0,04	1.81 ± 0.05	
8	137ah	> 2,50	> 2,50	> 2,50	> 2,50	> 20	
Ellipticin		1.15 ± 0.01	$1,51\pm0,05$	$1,60 \pm 0,02$	$1,82\pm0,03$	$6{,}18\pm0{,}11$	

Table 3.16. Cytotoxicity of compounds 137 qx

The results showed that most of synthesized podophyllotoxinnaphthoquinone compounds 137 exhibited inhibitory effect on cancer cells tested at different concentrations, 26/31 compounds displayed high potent inhibitory activities with $IC_{50} < 2,5 \mu M$. In general, podophyllotoxin-naphthoquinones **137a-x**, **127aa-ah** had less cytotoxic inhibitory activity against MCF7 cancer cell line than other cancer cell lines with $IC_{50} > 2,50 \mu M$. Many synthetic podophyllotoxin-naphthoquinone compounds were shown to have higher inhibitory activity than these of the reference ellipticine. In particular, compounds **137j** (Ar = 3-MeOC₆H₄), **137k** (Ar = 3-BrC₆H₄), **137l** (Ar = 3-NO₂C₆H₄), **137m** (Ar = C₆H₅) and **137af** (Ar = 3,4,5-(MeO)₃C₆H₂) were found to be the most potent antineoplastic agents with $IC_{50} < 40 \text{ nM}$ against HepG2 cells, SK- Lu-1 and KB. Compounds **137h** (Ar = 1-naphthyl), **137s** (Ar = 2,6-F₂-4-MeOC₆H₂), **137t** (Ar = 2-CF₃-4-MeOC₆H₃), **137ah** (Ar = 5-Br-pyridin-2-yl) were less cytotoxic to all 4 cancer cell lines than other compounds with $IC_{50} > 2,50 \mu M$. In addition, all compounds **137a-ah** (except **137af**) exhibited low toxicity or moderate to Hek-293 cell line with IC_{50} more than 1,08 μ M, even many compounds more than 20 μ M.

In addition, the most potent compounds **137j** and **137k** were investigated further to assess their ability to influence cell cycle progression and apoptosis in SK-Lu-1 cells. The results confirmed that tested compounds possessed anti-proliferative activity through concentration-dependently inducing a significant G2/M-phase arrest, which is a representative feature shared by tubulin polymerization inhibitors. In addition, compounds **137j** and **137k** enhanced 4-5-fold caspase-3/7 activity as compared to Ellipticine. Tested with Annexin V and PI/dead cell apoptosis[®], compounds **137j**,**k** exhibited the anti-proliferative effect through dose-dependently triggering cellular apoptosis of SK-Lu-1 cells, particularly in the early apoptotic stage.

Additionally, molecular docking studies were performed and showed importance interaction of two compounds against residues in the colchicine-binding-site of tubulin as well. Compounds **137j** and **137k** exhibited activity as a classical tubulin inhibitor, similarly interacted with residues at the zone 1 and 2 of CBS and provided lower affinity against tubulin target than colchicine.

CONCLUSIONS

1. Developed a new process for successfully synthesizing 23 3benzoyl-4*H*-benzo[*g*]chromene-5,10-dione compounds **135a-w** in 60-88% yields through a three-component domino reaction as well as proposed a reaction mechanism to form products from 2-hydroxy-1,4-naphthoquinone, arylenaminone and aldehydes.

2. Developed a new process for successfully synthesizing 31 podophyllotoxin-naphthoquinone compounds **137a-x**, **137aa-ah** in 69-89% yields through a multicomponent domino reaction as well as proposed a reaction mechanism to form products from 2-hydroxy-1,4-naphthoquinone or 2-amino-1,4-naphthoquinone, tetronic acid and aldehydes.

3. Developed a new process for successfully synthesizing 17 *N*-arylated-dihydrobenzo[g]quinoline-5,10-diones **140a–q** in 69-86% yields through a multicomponent domino reaction and proposed reaction mechanism to form products from 2-hydroxy-1,4-naphthoquinone, aniline/aniline derivatives, tetronic acid and aldehydes.

4. Developed a new process for successfully synthesizing 9 benzo[*a*]pyridazino[3,4-*c*]phenazines **143a-i** 41-63% yields through a fourcomponent domino reaction and proposed a reaction mechanism to form products from 2-hydroxy-1,4-naphthoquinone, methyl hydrazine and ophenylenediamine and aldehydes.

5. The cytotoxic activity of 57 synthesized compounds was evaluated against cancer cell lines including epidermoid carcinoma (KB), hepatoma carcinoma (HepG2), non-small lung (SK-Lu-1 or A549) and breast (MCF7) cancer cell lines. The results showed that 43 compounds exhibited cytotoxicity against tested cancer cell lines, of which 25 compounds showed higher cytotoxic activity than those of the reference ellipticine. Compounds **137i**, **137j**, **137k**, **137l**, **137af** displayed the highest cytotoxic activity with IC₅₀ < 50 nM.

6. The initial findings have indicated the anticancer mechanism of podophyllotoxin-naphthoquinone compounds **137j** and **137k** by arresting the G2/M phase of the cell cycle, activating caspase-3/7 enzymes, and inducing apoptosis. Additionally, the tubulin inhibition mechanism of these two compounds has also been proposed through molecular docking simulations.

NEW CONTRIBUTIONS OF THE THESIS

1. Developed 4 new processes for the successful synthesis of quinonefused heterocyclic derivatives through multicomponent domino reactions.

2. Synthesized 80 new quinone-fused heterocyclic derivatives, which have not been published in previous documents, including:

- 23 3-Benzoyl-4*H*-benzo[g]chromene-5,10-dione compounds 135a-w.

- 31 Podophyllotoxin-naphthoquinone compounds 137a-x, 137aa-ah.

- 17 *N*-Arylated-dihydrobenzo[g]quinoline-5,10-dione compounds **140a–q**.

- 09 Benzo[a]pyridazino[3,4-c]phenazine compounds 143a-i.

3. The cytotoxic activity of 57 synthesized compounds was evaluated against cancer cell lines including epidermoid carcinoma (KB), hepatoma carcinoma (HepG2), non-small lung (SK-Lu-1 or A549) and breast (MCF7) cancer cell lines. The results showed that 43 compounds exhibited cytotoxicity against tested cancer cell lines, of which 25 compounds showed higher cytotoxic activity than those of the reference ellipticine. Compounds **137i**, **137j**, **137k**, **137l**, **137af** displayed the highest cytotoxic activity with IC₅₀ < 50 nM.

4. The initial findings have indicated the anticancer mechanism of podophyllotoxin-naphthoquinone compounds **137j** and **137k** by arresting the G2/M phase of the cell cycle, activating caspase-3/7 enzymes, and

inducing apoptosis. Additionally, the tubulin inhibition mechanism of these two compounds has also been proposed through molecular docking simulations.

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